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Synthesis of a library of benzoindolizines using poly(ethylene glycol) as soluble support

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Abstract—A library of benzoindolizines (pyrrolo [1,5-a] quinolines 10 and pyrrolo [1,5-a] quinolines 9) has been synthesized using poly(ethylene glycol) (PEG) as soluble polymer support. The PEG-supported isoquinolinium salt 4 reacted, respectively, with active alkenes 11 using tetrakispyridinecobalt(II) dichromate (TPCD) as oxidant or alkynes 12 to give 10, of which yields were from moderate to high. By analogy, the reaction of PEG-supported quinolinium salt 3 with 12 was to produce 9. However, in the presence of TPCD the reaction of 3 with 11 afforded indolizines 8, which was discovered firstly.

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1. Introduction

Indolizine is an important ring system in view of its similarity to indole. Recently many papers have described the application of indolizine derivatives in biology and medicine. They have not only been used as novel potent inhibitor of 15-lipoxyenase, ¹ calcium entry blockers, ² antileishmanial and antiviral agent, ³ histamine H_3 receptor antagonists, ⁴ but also have shown antimycobacterial activity, ⁵ antioxidant properties ⁶ and delayed replicative senescence of human diploid fibroblasts. ⁷ In addition, (\pm)-monomorine, ⁸ new class of fluorescent β -cyclodextrins ⁹ and di- or tetrahydroxyindolizidine ¹⁰ have been synthesized using them by miscellaneous methods.

In principle, there may be three approaches to synthesize indolizine. The most general method is the formation of the five-membered ring moiety in the indolizine framework mainly by intra- or intermolecular condensation, 4 1,3-dipolar cycloadditions, 11 1,5-dipolar cyclization. 6,12 Recently some other methods constructing five-membered rings have been reported such as Cu-assisted cycloisomerization of alkynyl imines, 8 electrophilic Baylis–Hillman reaction, 13 one-step gas-phase synthesis, 14 Aza Wolff rearrangement for 4-oxoquinolizine-3-diazonium tetrafluoroborates, 15 cerium(III)-catalyzed cycliza-

Keywords: Poly(ethylene glycol); Polymer-supported; Ylide; Benzoindolizine.

tion with 1-cyanomethylene tetrahydro-isoquionline, ¹⁶ the reaction of 2-formyl-1,4-dihydropyridine derivative with malonitrile ¹⁷ or activated methylene reagent. ¹⁸ The formation of six-membered rings by electrochemically induced hetero-[4 + 2]-cycloadditon, ¹⁹ condensation by BF₃(OEt₂) as catalyst ²⁰ and intramolecular Dieckmann type cyclization ²¹ may be the second type of approach for which there are only a few examples. The third method is the simultaneous formation of the five-membered and six-membered rings, which is only an example recently. ²² There are many methods introduced, however, the most important methods for the synthesis of indolizine and benzoindolizine derivatives is based on 1,3-dipolar cycloaddition reactions of Nheterocyclic ylides with electron-deficient alkenes and alkynes.

Currently, liquid-phase synthesis using soluble polymers technique greatly attracts the interests of organic chemists in the field of combinatorial chemistry. It has the advantages of conventional liquid-phase synthesis and easy separation purification of the products in solid-phase synthesis. Moreover, the soluble polymer-bound species allow using routine analytical methods (NMR, TLC or IR) to monitor the reaction process and to determine the structures of products attached to polymer support directly. Poly(ethylene glycol) (PEG) is an ideal support and the most widely used polymer for iquid-phase combinatorial synthesis in terms of its controllable solubility in different solvents. Therefore, we have reported that indolizines were synthesized effectively by PEG-supported pyridinium ylide reacted

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OH
$$\xrightarrow{a}$$
 OH \xrightarrow{a} OH \xrightarrow{b} OH \xrightarrow{a} OH \xrightarrow

Scheme 1. (a) BrCH₂COBr, DIPEA, CH₂Cl₂, 0°C. (b) Pyridine, rt CH₂Cl₂, overnight. (c) Alkenes 11, DIPEA, TPCD, DMF, 85-90°C, 3–4h. (d) Alkynes 12, DIPEA, toluene, 90°C, 3h. (e) 1% KCN/CH₃OH, rt 24h.

$$R^2$$
 R^1
 $MeOOC$
 $COOMe$
 R^3
 R^4
 R^4

Figure 1.

with active alkenes using TPCD as oxidant and alkynes (Scheme 1).²³

In an extension of our work on PEG-supported N-heterocyclic ylides, we here reported firstly the synthesis of new benzoindolizines (pyrrolo [1,5-a] quinolines 9 and pyrrolo [1,5-a] isoquinolines 10) by 1,3-dipolar cycloaddition of PEG-supported quinolinium and isoquinolinium ylides, respectively, with active alkenes 11 and alkynes 12. The structures of 11 and 12 were shown in Figure 1.

2. Results and discussion

At the beginning, we employed PEG-supported quinolinium ylide derived from quinolinium salt 3^{24} to react with 11 using TPCD (1 equiv) as oxidant and DIPEA (2 equiv) as base at 90 °C; for 3–4h, which did not get our anticipative pyrrolo [1,5-a] quinolines 9 but indolizines 8 (except that the reaction of 3 with 111 afforded the product 81 and 91 from ¹H NMR spectra) (Scheme 2). However, Zhang et al.²⁵ reported that the reaction of quinolinium ylides and, α , β -unsaturated ketones gave pyrrolo [1,5-a] quinolines. Under the above reaction condition, some yields of 8 were low. Therefore, optimizing the cycloaddition reaction conditions to obtain

Table 1. Synthesis of indolizines **8** from pyridinium ylide and quinolinium ylide with alkenes **11**

Entry	\mathbb{R}^1	\mathbb{R}^2	Yield	d ^a (%)
8a	COPh	Ph	_	50°
8b	COPh	p-Tolyl	_	52 ^c
8c	COPh	p-Anisyl	_	39 ^c
8d	COPh	<i>p</i> -FPh	_	37 ^c
8e	COPh	<i>p</i> -ClPh	_	29 ^c
8f	COPh	<i>p</i> -BrPh	_	31°
8g	COPh	p-NCPh	_	43°
8h	COPh	p-O ₂ NPh	_	67 ^c
8i	COPh	<i>m</i> -Furan	_	21°
8j	COMe	<i>m</i> -Furan	_	25°
8k	COMe	Ph	_	$30^{\rm c}$
81	COOMe	Н	_	40^{cd}
8m	COOMe	COOMe	75 ^b	34 ^c
8n	CN	Н	82 ^b	60^{c}
80	o,m-Cyclohe	xone	70 ^b	25°

^a Based on the loading capacity of PEG and purified by by column chromatography.

better yield, we choose K₂CO₃ to replace DIPEA and prolonged the reaction time, of which yields were improved (Table 1). Whereas, enhancing the reaction temperature did not improve the yields. It was obvious that the yields were less than those of 8a–o synthesized by the reaction of PEG-bound pyridinium salt with 11 in the presence of TPCD. The structure of 8m which was obtained by this method was confirmed by MS, ¹H and ¹³C NMR, ¹H–¹H COSY and DEPT spectral data. Because of giving products of indolizines, we considered if the oxidant TPCD was excessive. We tried decreasing TPCD or changing the reaction temperature, but the product were still 8²⁶ (except for 11l which still produce 8l and 9l).

The previous papers¹¹ reporting indolizine derivatives seldom or did not obtain 9 by 1,3-dipolar cycloaddition reaction of quinolinium ylides. It is difficult to explain the mechanism of the oxidation course. In addition, we employed MnO₂ or CrO₃ to replace TPCD in the reaction of 11m and found that using CrO₃, no product obtained and the product 9m was afforded in the presence of MnO₂ (the yield was very low). Therefore, the results implied that quinoline moiety were oxidated to pyridine ring by TPCD in the reaction.

Scheme 2. (a) Quinoline, rt CH₂Cl₂, overnight. (b) Alkenes 11, 1 equiv K₂CO₃, 1 equiv TPCD, DMF, 90 °C, overnight. (c) 1% KCN/CH₃OH, rt 24h.

^b From pyridinium salt 2, 8a-l see Ref. 23.

^c From quinolinium salt 3 in the presence of 1 equiv TPCD.

d Containing 81 (main) and 91.

Scheme 3. (a) Alkynes 12, 1 equiv K₂CO₃, DMF, 90 °C, overnight. (b) 1% KCN/CH₃OH, rt 24h.

Table 2. Synthesis of pyrrolo [1,5-a] quinolines 9 from alkynes 12

Entry	\mathbb{R}^3	R^4	Yielda (%)
91	COOMe	Н	51
9m	COOMe	COOMe	55

^a Based on the loading capacity of PEG and purified by column chromatography.

In succession, treatment of 3 with alkynes 12 using K_2CO_3 as base at 90 °C overnight obtained PEG-bound pyrrolo [1, 5-a] quinolines 6 and cleaved, purified by column chromatography to give anticipative pyrrolo [1,5-a] quinolines 9^{27} (Scheme 3). The yields were given in Table 2.

The annelation reaction was also extended to PEG-bound isoquinolinium salts 4²⁸ derived from isoquinoline (Scheme 4). Treatment of 4, respectively, with 11 and 12 finally afforded the corresponding products 10,²⁹ in moderate to high yields (Tables 3 and 4). In this reaction, the 8 or isomer products did not occur. The structure of 10a–o were confirmed by element analyses, IR, ¹H NMR spectral data (10c,e,l,n were given ¹³C NMR spectra).

In addition we further confirmed the structure of **10c** (Fig. 2) by MS, $^{1}H^{-1}H$ COSY, NOE 1D, $^{1}H^{-13}C$ COSY and DEPT spectra data.

3. Conclusion

In conclusion, we have developed a novel and efficient liquid-phase synthesis of pyrrolo [1,5-a] quinolines and pyrrolo [1,5-a] isoquinolines by a one-pot reaction via 1,3-dipolar cycloaddition by the soluble PEG-supported

Table 3. Synthesis of pyrrolo [1,5-a] isoquinolines 10 from alkenes 11

Entry	\mathbb{R}^1	\mathbb{R}^2	Yield ^a (%)
10a	COPh	Ph	71
10b	COPh	<i>p</i> -Tolyl	73
10c	COPh	p-Anisyl	84
10d	COPh	p-FPh	79
10e	COPh	p-ClPh	68
10f	COPh	<i>p</i> -BrPh	62
10g	COPh	p-NCPh	74
10h	COPh	p-O ₂ NPh	75
10i	COPh	<i>m</i> -Furan	56
10j	COMe	<i>m</i> -Furan	50
10k	COMe	Ph	58
10l	COOMe	Н	70
10m	COOMe	COOMe	72
10n	CN	Н	80
10o	o,m-cyclohexo	one	65

^a Based on the loading capacity of PEG and purified by by column chromatography.

Table 4. Synthesis of pyrrolo [1,5-a] isoquinolines 10 from alkynes 12

Entry	R ³	R ⁴	Yield ^a (%)
101	COOMe	Н	79
10m	COOMe	COOMe	82

^a Based on the loading capacity of PEG and purified by column chromatography.

Figure 2.

O Br a O
$$\frac{3}{2}$$
 $\frac{4}{5}$ $\frac{5}{6}$ $\frac{b}{7}$ $\frac{6}{c}$ $\frac{b}{R^{1(3)}}$ $\frac{1}{7}$ $\frac{6}{R^{2(4)}}$ $\frac{1}{10}$ $\frac{8}{10}$ $\frac{7}{10}$ $\frac{6}{10}$ $\frac{5}{10}$ $\frac{1}{10}$ $\frac{1}{2}$ $\frac{1}{10}$ $\frac{1}{2}$ $\frac{1}$

Scheme 4. (a) Isoquinoline, rt CH₂Cl₂. (b) Alkenes 11, 1 equiv DIPEA, 1 equiv TPCD, DMF, 90 °C, 3–4 h. (c) Alkynes 12, DIPEA, toluene, 90 °C, 3h. (d) 1% KCN/CH₃OH, rt 24h.

salts. Furthermore, firstly we discovered that PEG-supported quinolinium salt reacted with active alkenes in the presence of oxidant TPCD afforded indolizines, but not pyrrolo [1,5-a] quinolines.

4. Experimental

4.1. General preparation of PEG-supported salts 3 or 4

To a solution of PEG₃₄₀₀ (8 g, 4.71 mmol OH) and DI-PEA (1.6 mL, 9.42 mmol) in dry CH₂Cl₂ (50 mL) was added dropwise a solution of bromoacetyl bromide (0.8 mL, 9.42 mmol) in dry CH₂Cl₂ (2 mL) at 0 °C and stirred at rt overnight. The mixture was washed with water to remove ammonium bromide, dried with over MgSO₄ and concentrated. After precipitation with cold Et₂O, washing with cold Et₂O and drying under vacuum, a white solid 1 was obtained. Into a solution of 1 in dry CH₂Cl₂ (20 mL) was added quinoline or isoquinoline (1.82g, 14.13 mmol) and stirred at rt for 18h. After precipitation from cold Et₂O, the suspension was filtered and washed with cold Et₂O to give solid 3 (95%) and 4 (97%), respectively. TLC (EtOAc-petroleum ether, 1:4) showed that the solid is free from any low molecular reactants and by-products.

4.2. Typical procedure for preparation of 8 or 10 from 11

A solution of 3 or 4 (2g, 0.51 mmol), 11 (3.06 mmol), TPCD (0.62g, 1.02 mmol) and K_2CO_3 (0.28 g, 2.04 mmol) or DIPEA (0.35 mL, 2.04 mmol) in DMF (30 mL) was stirred at 80–90 °C for overnight or 3–4h. After the solvent was evaporated under vacuum, the residue was added CH_2Cl_2 (50 mL) and filtered. The filtrate was washed with water to remove ammonium bromide, dried with over MgSO₄, filtered, concentrated and precipitated with cold Et_2O to give 5 or 7. Product 5 or 7 was treated with a 1% solution of KCN in CH_3OH (30 mL) and stirred at rt overnight, evaporated CH_3OH and precipitated with cold Et_2O to give the crude products, which were purified by column chromatography on silica gel (EtOAc–petroleum ether, 1:4–1:2) to afford the pure 8 or 10.

4.3. Typical procedure for preparation of 9 or 10 from 12

A solution of 3 or 4 (2g, 0.51 mmol), 12 (3.06 mmol) and K_2CO_3 (0.28g, 2.04 mmol) or DIPEA (0.35 mL, 2.04 mmol) in toluene was stirred at 90 °C for overnight or 3 h. After solvent was removed, the residue was dissolved in CH_2Cl_2 (50 mL) washed, dried, filtered, concentrated, precipitated and cleaved by using the procedure described above for 8 and 10. The crude product was purified by column chromatography on silica gel (EtOAc–petroleum ether, 1:3–1:2) to afford pure 9 or 10.

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References and notes

- Gundersen, L.-L.; Malterud, K. E.; Negussie, A. H.; Rise, F. R.; Teklu, S.; Ostby, O. B. *Bioorg. Med. Chem.* 2003, 11, 5409.
- Gupta, S. P.; Mathur, A. N.; Nagappa, A. N.; Kumar, D.; Kumaran, S. Eur. J. Med. Chem. 2003, 38, 867.
- (a) Medda, S.; Jaisankar, P.; Manna, R. K.; Pal, B.; Giri, V. S.; Basu, M. K. J. Drug Target 2003, 11, 123; (b) Bolle, L. D.; Andrei, G.; Snoeck, R.; Zhang, Y.; Lommel, A. V.; Otto, M.; Bousseau, A.; Roy, C.; Clercq, E. D.; Naesens, L. Biochem. Pharmacol. 2004, 67, 325.
- 4. Chai, W.; Breitenbucher, J. G.; Kwok, A.; Li, X.; Wong, V.; Carruthers, N. I.; Lovenberg, T. W.; Mazur, C.; Wilson, S. J.; Axe, F. U.; Jones, T. K. *Bioorg. Med. Chem. Lett.* **2003**, *13*, 1767.
- Gundersen, L.-L.; Negussie, A. H.; Ostbly, O. B. Arch. Pharm. Pharm. Med. Chem. 2003, 336, 191.
- Ostbly, O. B.; Dalhus, B.; Gundersen, L.-L.; Rise, F.; Bast, A.; Haenen, G. R. M. M. Eur. J. Org. Chem. 2000, 3763
- Wang, P.; Zhang, Z.; Ma, X.; Huang, Y.; Liu, X.; Tu, P.; Tong, T. Mech. Ageing Dev. 2003, 124, 1025.
- Kel'in, A. V.; Sromek, A. W.; Gevorgyan, V. J. Am. Chem. Soc. 2001, 123, 2074.
- Delattre, F.; Woisel, P.; Surpateanu, G.; Bria, M.; Cazier, F.; Decock, P. Tetrahedron 2004, 60, 1557.
- Paolucci, C.; Mattioli, L. J. Org. Chem. 2001, 66, 4787.
- Please see Ref. 23, others: pyridinium ylide (a) Fang, X.; Wu, Y.-M.; Deng, J.; Wang, S.-W. Tetrahedron 2004, 60, 5487; (b) Komatsu, M.; Kasano, Y.; Yamaoka, S.; Minakata, S. Synlett 2003, 1398; (c) Ghosh, C. K.; Karak, S. K.; Patra, A. Heterocycles 2003, 60, 825; (d) Georgescu, E.; Georgescu, F.; Danila, M.; Filip, P. I.; Draghici, C.; Caproiu, M. T. Arkivoc 2002, ii, 30; (e) Matsumoto, K.; Tanaka, N.; Uchida, T.; Ikemi, Y.; Hayashi, N.; Aoyama, K.; Kakehi, A. Heterocycles 2000, 54, 611; (f) Dinica, R. M.; Druta, I. I.; Druta, I. I.; Pettinari, C. Synlett 2000, 1013; (g) Andrei, M. A.; Ganj, C.; Aburel, P. S. Tetrahedron 1999, 55, 13063; (h) azomethine ylide: Bashiardes, G.; Safir, I.; Barbot, F.; Laduranty, J. Tetrahedron Lett. 2004, 45, 1567; (i) Pearson, W. H.; Stoy, P.; Mi, Y. J. Org. Chem. 2004, 69, 1919.
- (a) Kakehi, A.; Suga, H.; Kako, T.; Fujii, T.; Tanaka, N.; Kobayashi, T. Chem. Pharm. Bull. 2003, 51, 1246; (b) Kakehi, A.; Ito, S.; Suga, H.; Miwa, T.; Mori, T.; Fujii, T.; Tanaka, N.; Kobayashi, T. Chem. Pharm. Bull. 2003, 51, 75; (c) Kakehi, A.; Ito, S.; Suga, H.; Miwa, T.; Mori, T.; Kobayashi, T. Heterocycles 2002, 57, 17; (d) Kakehi, A.; Ito, S.; Suga, H.; Miwa, T.; Mori, T.; Kobayashi, T. Heterocycles 2001, 57, 17; (e) Kakeshi, A.; Ito, S.; Suga, H.; Yasuraoka, K. Heterocycles 2001, 54, 185; (f) Kakehi, A.; Ito, S.; Hirata, K.; Zuo, P. Chem. Pharm. Bull. 2000, 48, 865.
- 13. Basavaiah, D.; Rao, A. J. Chem. Commun. 2003, 604.
- Sparrapan, R.; Mendes, M. A.; Carvalho, M.; Eberlin, M. N. Chem. Eur. J. 2000, 6, 321.
- Recnik, S.; Svete, J.; Stanovnik, B. Eur. J. Org. Chem. 2001, 3705.
- Vincze, Z.; Nemes, P.; Balazs, B.; Toth, G.; Scheiber, P. Synlett 2004, 1023.
- 17. Marchalin, S.; Cvopova, K.; Pham-Huu, D.-P.; Chudik, M.; Kozisek, J.; Svoboda, I.; Daich, A. *Tetrahedron Lett.* **2001**, *42*, 5663.
- 18. Marchalin, S.; Chudik, M.; Cvopova, K.; Kozisek, J.; Lesko, J.; Daich, A. Tetrahedron 2002, 58, 5747.
- Peglow, T.; Blechert, S.; Steckhan, E. Chem. Eur. J. 1998, 4, 107.

- Kobayashi, K.; Nakahashi, R.; Takanohashi, A.; Kitamura, T.; Morikawa, O.; Konishi, H. Chem. Lett. 2002, 624.
- Ong, C. W.; Lai, M. C.; Ian, J. J.; Chang, Y. A. Heterocycles 2002, 57, 1303.
- 22. Fujii, H.; Iwafuji, K.; Sawae, Y.; Abe, N.; Kakehi, A. *Heterocycles* **2003**, *59*, 63.
- 23. Chen, Z.; Yue, G.; Lu, C.; Yang, G. Synlett 2004, 1231.
- 24. PEG-quinolinium salt 3: IR (KBr): 2884, 1750, 1476, 1280, 1114, 843 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): δ = 10.47 (d, 1H, J = 6.0 Hz, H-2), 9.10 (d, 1H, J = 8.4 Hz, H-4), 8.57 (d, 1H, J = 8.4 Hz, H-8), 8.30 (d, 1H, J = 8.4 Hz, H-5), 8.19 (t, 1H, J = 7.8 Hz, H-7), 8.01 (t, 1H, J = 7.8 Hz, H-6) 7.92 (t, 1H, J = 7.6 Hz, H-3), 6.55 (s, 2H, -CH₂CO-), 3.89–3.48 (m, 4n H, -O(CH₂ CH₂O)_n-).
- Zhang, X.; Cao, W.; Wei, X.; Hu, H. Synth. Commun. 1997, 27, 1395.
- 26. **8a**-m: ¹H NMR (600 MHz, CDCl₃) see Ref. 23. **8m**: ¹³C NMR (150 MHz, CDCl₃): $\delta = 166.28$, 163.26, 160.34, 137.73, 130.47, 127.83, 126.76, 119.82, 115.42, 111.70, 115.42, 111.70, 102.89, 52.88, 51.96, 51.62. MS (ESI) (m/ z): 260 (100), 292 (81) $[M + H]^+$, 314 (59) $[M + 23]^+$, 330 (11) [M + 39]⁺. 8n: IR (KBr): 2217, 1694, 1520, 1491, 1445, $763 \,\mathrm{cm}^{-1}$. ¹H NMR (600 MHz, CDCl₃): $\delta = 9.53$ (d, 1H, J = 7.2 Hz, H-5, 7.782 - 7.766 (sd, 2H, H-8, H-2), 7.37 (t,1H, J = 7.8, 8.4 Hz, H-7), 7.06 (t, 1H, J = 7.2, 6.6 Hz, H-6), 3.94 (s, 3H, COOMe). ¹³C NMR (150 MHz, CDCl₃): $\delta = 60.71$, 140.58, 128.24, 125.88, 124.78, 117.63, 115.36, 115.12, 115.00, 83.85, 51.67. Anal. Calcd for C₁₁H₈N₂O₂: C, 63.44; H, 7.74; N, 13.45. Found: C, 63.45; H, 4.00; N, 13.47. **80**: IR (KBr): 2948, 1688, 1649, 1510, 1438, 780 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): δ = 9.61 (d, 1H, J = 7.8 Hz, H-5, 8.45 (d, 1H, J = 7.2 Hz, H-8), 7.41 (t, 1H, $J = 7.8, 7.2 \,\text{Hz}, \text{H}-7$, 7.05 (t, 1H, $J = 6.6 \,\text{Hz}, \text{H}-6$), 3.95 (s, 3H, COOMe), 3.21 (t, 2H, J = 6.3 Hz, COCH₂), 2.62 (t, 2H, J = 6.3 Hz, CH₂), 2.16 (m, 2H, CH₂). Anal. Calcd for C₁₄H₁₃NO₃: C, 69.12; H, 5.39; N, 5.76. Found: C, 69.10; H, 5.35; N, 5.77.
- 27. Pyrrolo [1,5-a] quinolines 9: 91: IR (KBr): 1704, 1615, 1551, 1432, 1081, 813, 754cm⁻¹. ¹H NMR (600 MHz, CDCl₃): $\delta = 8.43$ (d, 1H, J = 9.0 Hz, Ar-H), 8.31 (d, 1H, $J = 9.6 \,\mathrm{Hz}$, Ar-H), 7.99 (s, 1H, H-2), 7.80 (d, 1H, $J = 7.2 \,\mathrm{Hz}$, Ar-H), 7.63 (t, 1H, $J = 9.6 \,\mathrm{Hz}$, Ar-H), 7.50 (t, 1H, J = 7.50 Hz, Ar-H), 3.98 (s, 3H, COOMe), 3.93 (s, 3H, COOMe). ¹³C NMR (150 MHz, CDCl₃): $\delta = 164.46$, 162.17, 139.45, 134.55, 133.34, 128.61, 128.33, 127.87, 126.57, 125.36, 125.15, 119.98, 117.64, 107.10, 52.11, 51.32. Anal. Calcd for C₁₆H₁₃NO₄: C, 71.90; H, 4.90; N, 5.24. Found: C, 71.87; H, 4.96; N, 5.18. 9m: IR (KBr): 1706, 1612, 1546, 1482, 1448, 814, 757, 742 cm⁻¹. ¹H NMR (600 MHz, CDCl₃) and ${}^{1}H^{-1}H$ COSY: $\delta = 8.25$ (d, 1H, J = 9.6 Hz, H-5), 8.10 (d, 1H, J = 9.0 Hz, H-10), 7.80 (d, 1H, J = 7.2 Hz, H-7), 7.65-7.60 (m, 2H, H-6, H-9), 7.52(t, 1H, J = 7.5 Hz, H-8), 4.00 (s, 3H, COOMe), 3.97 (s, 3H, COOMe)COOMe), 3.92 (s, 3H, COOMe). ¹³C NMR (150 Hz, CDCl₃): δ = 166.16, 163.34, 161.20, 137.86, 132.76, 130.99, 128.85, 128.66, 128.61, 125.86, 125.33, 119.52, 117.64, 117.59, 105.18, 52.85, 52.52, 51.81. Anal. Calcd for C₁₈H₁₅NO₆: C, 63.34; H, 4.43; N, 4.10. Found: C, 63.38; H, 4.40; N, 4.17.
- 28. PEG-isoquinolinium salt **4**: IR (KBr): 2884, 1750, 1467, 1113, 947, 842 cm $^{-1}$. 1 H NMR (600 MHz, CDCl₃): δ = 10.87 (s, 1H, H-1), 8.77 (d, 1H, J = 5.4 Hz, H-3), 8.63 (d, 1H, J = 8.4 Hz, H-8), 8.40 (d, 1H, J = 6.0 Hz, H-4), 8.23 (d, 1H, J = 7.8 Hz, H-5), 8.18 (t, 1H, J = 7.5 Hz, H-6), 7.99 (t, 1H, J = 7.5 Hz, H-7), 6.28 (s, 2H, -CH₂CO-), 3.77–3.52 (m, 4n H, -O(CH₂CH₂O)_n-).
- 29. Pyrrolo [1,5-*a*] isoquinolines **10**: **10a**: IR (KBr): 1693, 1658, 1598, 796, 732, 669 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): δ = 9.41 (d, 1H, J = 7.8 Hz, Ar-H), 7.95 (d, 1H,

J = 8.4Hz, Ar-H), 7.74 (d, 2H, J = 7.2Hz, Ar-H), 7.70 (d, 1H, J = 9.0 Hz, Ar-H), 7.48(t, 1H, J = 7.5 Hz, Ar-H), 7.41 (t, 1H, J = 7.2 Hz, Ar-H), 7.34 (t, 1H, J = 7.8 Hz, Ar-H), 7.26-7.13 (m, 8H, Ar-H), 3.67 (s, 3H, COOMe). Anal. Calcd for C₂₇H₁₉NO₃: C, 79.98; H, 4.72; N, 3.45. Found: C, 80.05; H, 4.81; N, 3.50. 10b: IR (KBr): 1690, 1661, 1520, 1435, 796, 735, 691 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): $\delta = 9.39$ (d, 1H, J = 7.8 Hz, Ar-H), 7.90 (d, 1H, $J = 7.8 \,\mathrm{Hz}$, Ar-H), 7.76 (d, 2H, $J = 7.2 \,\mathrm{Hz}$, Ar-H), 7.69 (d, 1H, J = 7.8 Hz, Ar-H), 7.46–7.10 (m, 8H, Ar-H), 6.96 (d, 2H, J = 7.8 Hz, Ar-H), 3.67 (s, 3H, COOMe), 2.25 (s, 3H, Me). Anal. Calcd for C₂₈H₂₁NO₃: C, 80.17; H, 5.05; N, 3.34. Found: C, 80.11; H, 4.97; N, 3.35. 10c: IR (KBr): 1690, 1659, 1522, 1435, 796, 737, 691 cm⁻¹. ¹H NMR $(600 \,\mathrm{MHz}, \,\mathrm{CDCl_3})$: $\delta = 9.40 \,\mathrm{(d, 1H, } J = 7.2 \,\mathrm{Hz}, \,\mathrm{H-5}), \,7.92$ (d, 1H, J = 8.4 Hz, H-10), 7.75 (d, 2H, J = 7.8 Hz, H-16a, 16b), 7.69 (d, 1H, J = 7.8 Hz, H-7), 7.45 (t, 1H, J = 7.5 Hz, H-3), 7.40 (t, 1H, $J = 7.2 \,\text{Hz}$, H-11), 7.31 (t, 1H, $J = 7.8 \,\mathrm{Hz}, \, \mathrm{H}$ -9), 7.25 (t, 2H, $J = 7.5 \,\mathrm{Hz}, \, \mathrm{H}$ -17a, 17b), 7.14 (d, 2H, $J = 8.0 \,\text{Hz}$, H-20a, 20b), 7.12 (d, 1H, J = 7.2 Hz, H-6), 6.70 (d, 2H, J = 8.4, H-21a, 21b), 3.74 (s, 3H, COOMe), 3.68 (s, 3H, OMe). ¹³C NMR (150 MHz, CDCl₃): $\delta = 196.35$ (C-14), 162.23 (C-24), 158.64 (C-23), 138.40 (C-15), 135.61 (C-2), 133.14 (C-18), 131.41 (C-13), 131.36 (C-20a, b), 129.74 (C-16a, b), 129.03 (C-12), 128.25 (C-17a, b), 127.93 (C-9), 127.68 (C-8), 126.96 (C-7), 126.27 (C-19), 124.71 (C-10), 124.65 (C-5), 123.98 (C-3), 118.34 (C-6), 113.36 (C-1), 112.68 (C-21a, b), 55.04, (C-23) 51.04 (C-25). MS (ESI) (m/z): 404 (15), 436 (100) $[M + H]^+$, 458 (46) $[M + 23]^+$, 474 (6) $[M + 39]^+$. Anal. Calcd for $C_{28}H_{21}NO_4$: C, 77.23; H, 4.86; N, 3.22. Found: C, 77.36; H, 4.82; N, 3.29. 10d: IR (KBr): 1692, 1657, 1521, 1436, 797, 736, 691 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): δ = 9.41 (d, 1H, J = 7.2 Hz, Ar-H), 7.94 (d, 1H, J = 7.8 Hz, Ar-H), 7.74–7.70 (m, 3H, Ar-H), 7.49–7.17 (m, 9H, Ar-H), 6.86 (t, 1H, J = 8.4Hz, Ar-H) 3.66 (s, 3H, COOMe). MS (ESI) (m/z): 288 (94), 316 (43), 392 (12), 424 (100) $[M + H]^+$, 446 (48) [M + 23]⁺. Anal. Calcd for C₂₇H₁₈NFO₃: C, 76.59; H, 4.28; N, 3.31. Found: C, 76.46; H, 4.12; N, 3.38. 10e: IR (KBr): 1693, 1657, 1512, 1436, 794, 739, 709 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): $\delta = 9.40$ (d, 1H, J = 7.8 Hz, Ar-H), 7.91 (d, 1H, J = 8.4 Hz, Ar-H), 7.73 (d, 2H, J = 7.2 Hz, Ar-H), 7.70 (d, 1H, J = 7.8 Hz, Ar-H), 7.49–7.13 (m, 11H, Ar-H), 3.66 (s, 3H, COOMe). ¹³C NMR (150 MHz, CDCl₃): $\delta = 195.81$, 161.93, 138.34, 134.45, 133.38, 133.13, 131.81(2 C), 131.58, 130.35 (2 C), 129.70, 129.07, 128.41, 128.14, 127.84, 127.05, 124.73, 124.61, 123.92, 121.50, 118.20, 114.13, 113.39, 51.12. Anal. Calcd for C₂₇H₁₈NClO₃: C, 73.72; H, 4.12; N, 3.18. Found: C, 73.78; H, 4.20; N, 3.24. **10f**: IR (KBr): 1693, 1656, 1509, 1436, 795, 735, 692 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): δ = 9.40 (d, 1H, 7.8 Hz, Ar-H), 7.90 (d, 1H, J = 8.4 Hz, Ar-H), 7.73 (d, 2H, J = 7.2 Hz, Ar-H), 7.68 (d, 1H, J = 7.8 Hz, Ar-H), 7.49-7.26 (m, 7H, Ar-H), 7.17 (d, 1H, J = 7.2 Hz, Ar-H), 7.09 (d, 2H, J = 8.4 Hz, Ar-H), 3.67 (s, 1H, COOMe). Anal. Calcd for C27H18NBrO3: C, 66.95; H, 3.75; N, 2.18. Found: C: 66.99; H, 3.62; N, 2.10. 10g: IR (KBr): 2227, 1695, 1657, 1516, 1436, 797, 732, 692 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): δ = 9.41 (d, 1H, J = 7.2 Hz, Ar-H), 7.91 (d, 1H, 8.4 Hz, Ar-H), 7.73-7.71 (m, 3H, Ar-H), 7.52–7.45 (m, 4H, Ar-H), 7.37–7.33 (m, 6H, Ar-H), 7.28 (t, 1H, J = 7.8 Hz, Ar-H), 7.21 (d, 1H, J = 7.8 Hz, Ar-H), 3.65 (s, 3H, COOMe). Anal. Calcd for $C_{28}H_{18}N_2O_3$: C, 78.13; H, 4.21; N, 6.51. Found: C, 78.20; H, 4.17; N, 6.61. **10h**: IR (KBr): 1696, 1517, 1344, 797, 732, 697 cm⁻¹ ¹H NMR (600 MHz, CDCl₃): δ = 9.39 (d, 1H, J = 7.8 Hz, Ar-H), 8.02 (d, 1H, J = 9.0 Hz, Ar-H), 7.71–7.23 (m, 12H, Ar-H), 7.19 (d, 1H, J = 7.8 Hz, Ar-H), 3.467 (s, 3H, COOMe). Anal. Calcd for C₂₇H₁₈N₂O₅: C, 71.99; H, 4.03;

N, 6.22. Found: C, 71.96; H, 4.13; N, 6.15.10i: IR (KBr): 1694, 1662, 1527, 1450, 1360, 794, 731, 691 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): $\delta = 9.32$ (d, 1H, J = 7.8 Hz, Ar-H), 8.02 (d, 1H, J = 8.4 Hz, Ar-H), 7.81 (d, 2H, J = 7.2 Hz, Ar-H), 7.66 (d, 1H, J = 7.8 Hz, Ar-H), 7.48–7.29 (m, 5H, Ar-H,), 7.18 (t, 1H, J = 1.2 Hz, furan-H), 7.12 (d, 1H, $J = 7.8 \,\mathrm{Hz}$, Ar-H), 6.45 (d, 1H, $J = 3.0 \,\mathrm{Hz}$, furan-H), 6.41 (m, 1H, furan-H), 3.84 (s, 3H, COOMe). Anal. Calcd for C₂₅H₁₇NO₄: C, 75.94; H, 4.33; N, 3.54. Found: C, 75.99; H, 4.29; N, 3.55. 10i: IR (KBr): 1697, 1673, 1511, 1453, 801, 742, 597 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): δ = 9.29 (t, 1H, J = 8.1 Hz, Ar-H), 8.45 (d, 1H, J = 7.8 Hz, Ar-H), 7.67 (t, 1H, J = 6.0 Hz, Ar-H), 7.58–7.51 (m, 3H, Ar-H), furan-H), 7.12 (t, 1H, J = 6.6 Hz, Ar-H), 6.64 (d, 1H, $J = 3.0 \,\text{Hz}$, furan-H), 6.55 (m, 1H, furan-H), 3.83 (s, 3H, COOMe), 2.23 (s, 3H, COMe). Anal. Calcd for C₂₀H₁₅NO₄C, 72.06; H, 4.54; N, 4.20. Found: C, 72.01; H, 4.53; N, 4.12. 10k: IR (KBr): 1691, 1506, 1436, 796, 711, 667 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): δ = 9.33 (d, 1H, J = 7.8 Hz, Ar-H), 8.41 (d, 1H, J = 8.4 Hz, Ar-H), 7.68(d, 1H, J = 8.4 Hz, Ar-H), 7.53–7.33 (m, 7H, Ar-H), 7.12 (d, 1H, J = 7.2 Hz, Ar-H), 3.60 (s, 3H, COOMe), 2.04 (s, 3H, COMe). Anal. Calcd for C₂₂H₁₇NO₃C, 76.95; H, 4.99; N, 4.08. Found: C, 76.89; H, 4.87; N, 4.11. 10l: IR (KBr): 1693, 1533, 1430, 798, 758 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): δ = 9.83 (d, 1H, J = 7.8 Hz, Ar-H), 9.38 (d, 1H, $J = 7.8 \,\mathrm{Hz}, \,\mathrm{Ar\text{-}H}), \,8.05 \,\mathrm{(s, 1H, H\text{-}2)}, \,7.74 - 7.62 \,\mathrm{(m, 3H, Ar\text{-}4)}$ H), 7.20 (d, 1H, J = 7.8 Hz, Ar-H), 3.95 (s, 3H, COOMe),

3.94 (s, 3H, COOMe). ¹³C NMR (150MHz, CDCl₃): 165.11, 165.38, 136.01, 129.63, 128.87, 127.75, 127.71, 126.67, 125.36, 124.87, 124.23, 115.59, 115.03, 109.21, 51.73, 51.56. Anal. Calcd for C₁₆H₁₃NO₄: C, 67.84; H, 4.63; N, 4.94. Found: C, 67.89; H, 4.61; N, 4.91. 10m: IR (KBr): 1740, 1705, 1532, 1439, 803 cm⁻¹. ¹H NMR $(600 \text{ MHz}, \text{CDCl}_3)$: $\delta = 9.46 \text{ (m, 1H, Ar-H)}, 9.36 \text{ (d, 1H, }$ $J = 7.8 \,\mathrm{Hz}$, Ar-H), 7.74–7.62 (m, 3H, Ar-H), 7.3 (d, 1H, $J = 7.8 \,\mathrm{Hz}$, Ar-H), 3.98 (s, 3H, COOMe), 3.93 (s, 3H, COOMe), 3.93 (s, 3H, COOMe). Anal. Calcd for C₁₈H₁₅NO₆: C, 63.34; H, 4.43; N, 4.10. Found: C, 63.35; H, 4.30; N, 4.15. 10n: IR (KBr): 2215, 1702, 1702, 1535, 1418, 794, 740, 688 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): δ = 9.32 (d, 1H, J = 7.8 Hz, H-5), 8.96 (d, 1H, J = 8.4 Hz, H-10), 7.80–7.27 (m, 3H, H-7, 8, 9), 7.24 (d, 1H, J = 7.8 Hz, H-6), 3.96 (s, 3H, COOMe). ¹³C NMR (150 MHz, CDCl₃): 160.66, 137.03, 129.52, 128.91, 128.69, 127.19, 124.40, 124.30, 123.85, 123.68, 117.24, 116.90, 115.24, 85.18, 51.84. Anal. Calcd for C₁₅H₁₀NO₂: C, 67.26; H, 4.27; N, 5.93. Found: C, 67.30; H, 4.28; N, 5.93. **10o**: IR (KBr): 2941, 1689, 1654, 1525, 1436, 806, 747 cm⁻¹. ¹H NMR (600 MHz, CDCl₃): δ = 10.23 (m, 1H, Ar-H), 9.51 (d, 1H, J = 7.2 Hz, Ar-H), 7.74–7.65 (m, 3H, Ar-H), 7.25 (d, 1H, J = 7.8 Hz, Ar-H), 3.98 (s, 3H, COOMe), 3.27 (t, 1H, J = 6.3 Hz, CH₂), 2.73 (t, 1H, $J = 6.3 \,\mathrm{Hz}$, CH₂), 2.19 (m, 1H, CH₂). Anal. Calcd for C₁₈H₁₅NO₃C, 73.71; H, 5.15; N, 4.78. Found: C, 73.73; H, 5.12; N, 4.78.